



Swati Yadav *et al*, Int. Journal of Pharmaceutical Sciences and Medicine (IJPSM),  
Vol.8 Issue. 7, July- 2023, pg. 67-80

ISSN: 2519-9889

Impact Factor: 5.9

# FORMULATION AND EVALUATION OF TRANSDERMAL PATCH OF HYDROXYCHLOROQUINE SULPHATE

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DOI: 10.47760/ijpsm.2023.v08i07.007

## ABSTRACT:

The aim of the present study was to investigate the patch for transdermal Hydroxychloroquine delivery in an effort to mask the bitter taste when orally administered. Hydroxychloroquine has easily measurable outputs that are linked to increased renal Na<sup>+</sup> excretion. We thus monitored urinary Na<sup>+</sup> output in separate groups intravenously administered Hydroxychloroquine or topically applied patch. Transdermal therapeutic systems are high- tech patches that make treatment much more convenient and pleasant for patient .Instead of having to take lots of tablets, patients often only have to apply a new patch once a week In addition, patches from LTS also ensure a constant drug level for the entire duration of wear Symptoms upon getting up in the morning as a result of low drug levels during the night are avoided. There are five major advantages of transdermal patches above other systems like Improved well –being for patient, constant drug levels, fewer side effects, Drugs that cannot be absorbed via the gastrointestinal tract can now be used and because the substance is released evenly and directly into the blood stream, less of the substance may be required.

**Keywords:** Transdermal patch, Hydroxychloroquine sulphate, Malaria, topical Drug Delivery system, Blood stream

## 1. INTRODUCTION

Transdermal drug delivery system (TDDS), skin patch or skin patch is a medicated adhesive patch that is placed on the skin to deliver drug through the skin and to the systemic circulation at a predetermined rate over a prolonged period of time. Transdermal patches are delivered the drug through the skin in controlled and predetermined manner so as to increase the therapeutic efficacy of drug and reduced side effect of drug. For effective Transdermal drug delivery system, the drugs are easily ready to penetrate the skin and easily reach the target. Hydroxychloroquine Sulphate has a bad teste (bitter) so patient having difficulty in swallowing especially in children.



Hydroxychloroquine Sulphate when administered in patient with auditory complications, it may reduce the hearing capacity of those patient .and when administered in high dosage it affects retinal sytem<sup>1</sup>.

Hydroxychloroquine Sulphate when administered orally it affects the gastrointestinal system causes nausea, vomiting, diarrhoea, abdominal cramps. When we administered Hydroxychloroquine Sulphate as a transdermal drug delivery system, this system release the drug with a zero order controlled delivery, and when any complications occurred in any patient due to this drug termination of therapy is possible only with this dosage form transdermal drug delivery system is the safest and convenient dosage form. Self medication is possible with this dosage form, a very convenient dosage form for child patients, having difficulty in swallowing<sup>2</sup>.

## 2. MATERIALS AND METHOD

### 2.1 MATERIALS

Hydroxychloroquine Sulphate pure drug sample was generously gifted by Yarrow chem products Pvt. Ltd. Gelatin from Centre drug house (P) LTD. PVP from Avantor Performance Materials India Limited. Dibutyl Pthalate from Centre drug house (p) ltd. Methanol and water used were analytical grade.

### 2.2 METHOD

#### 2.2.1 Melting point determination:

Melting point of Hydroxychloroquine sulphate was determined using capillary method. In this method little amount of Hydroxychloroquine sulphate was filled in capillary after that the capillary was tied to a thermometer with the help of a rubber band. The thermometer with capillary was placed into Theil's tube which was previously filled with paraffin oil. The paraffin oil in the tube was heated until the drug melts. The temperature at which drug begins to melt was recorded.<sup>3</sup>

#### 2.2.2 Determination of wavelength using UV spectrophotometric analysis:

50mg of Hydroxychloroquine sulphate was weighed and dissolved into 50ml of water to



prepare a 1000 $\mu$ g/ml stock solution from which a 10 $\mu$ g/ml dilution was prepared. Baseline correction was performed using water and sample was run between 200-400nm wavelength ranges in spectrum mode.<sup>3</sup>

### **2.2.3 Preparation of calibration curves:**

The calibration curve of Hydroxychloroquine sulphate were prepared in distilled water, 7.4 pH phosphate buffer and 6.8 pH phosphate buffer by using Shimadzu 1800 UV visible spectrophotometer.

Accurately weighed 50mg of Hydroxychloroquine sulphate was transferred into a 50ml volumetric flask and the volume was made up with water to obtain a 1000 $\mu$ g/ml stock solution of Hydroxychloroquine sulphate. From the stock solution 1ml was taken and transferred into a 10ml volumetric flask and rest of the volume was made up with water to obtain a 100 $\mu$ g/ml of solution from which 1 to 10 $\mu$ g/ml dilutions were prepared. Same procedure was followed for distilled water, phosphate buffer 6.8, and phosphate buffer 7.4 to prepare calibration curve respectively.<sup>3</sup>

### **2.2.4 Determination of Partition coefficient:**

The Partition coefficient of Hydroxychloroquine sulphate was determined by taking 20ml of octanol in a separating funnel and saturating it with 20ml of phosphate buffer pH 7.4 for overnight by intermittent shaking for 4h, keeping the separating funnel undisturbed for 1h after that 10mg of Hydroxychloroquine sulphate was added into the separating funnel with moderate shaking for 24h. The two layers were separated and filtered through syringe filter and drug concentration was determined in both phases by UV-visible spectrophotometric method at 220 nm<sup>3</sup>.

### **2.2.5 Determination of solubility of Hydroxychloroquine sulphate in various medium:**

The solubility of Hydroxychloroquine sulphate in various medium was determined by shake flask method. In this method 2ml of each solvent was taken into a vial and an excess amount of Hydroxychloroquine sulphate was added. The vials were sealed properly and stirred for 10min. They were then kept on orbital flask shaker at 37°C for 24h. After solubilization of Hydroxychloroquine sulphate an extra amount of Hydroxychloroquine sulphate was added to the vials containing drug-solvent mixture. The process was repeated until saturation solubility

of Hydroxychloroquine sulphate, indicated by presence of un dissolved drug. The mixtures were then kept at room temperature for 24 h. and centrifuged using Remi 12C micro-centrifuge at 3000RPM for 15min. The supernatant were separated and diluted with respective solvents. The drug concentration was analyzed spectrophotometrically at 255nm using UV-visible spectrophotometer (Shimadzu-1800).<sup>3</sup>

### 2.2.6 Drug-excipient interaction study:

The compatibility of the drug was assessed by drug-excipient interaction study. The drug was mixed with various excipients in a 1:1 ratio in glass vials which were properly sealed and kept undisturbed at 40°C temperature for 14 days. After 14 days incompatibility was confirmed by TLC.<sup>4</sup>

### 2.3 Method of Preparing Transdermal Patches:

Method of preparation of TDDS was summarized by modifying the earlier reported methods. The patches were prepared by solvent casting method. The polymer (for example PVP/HPMC) was taken in a beaker with a minimum quantity of the solvent. Then 2/3rd of the solvent was mixed with the other polymers (for example PVA) and was added firstly with stirring at lower rpm and later at a higher speed. The plasticizer was added and homogeneously mixed and the drug was included with enduring agitation and the volume was made up. The films were cast onto a suitably designed and fabricated glass mould and then dried in oven at 40 °C. The films were removed by using sharp blade by inserting along the edges of the film. The dried films were wrapped in butter paper and stored in a closed container away from light and in cool place.<sup>4</sup>

**Table 1: Formulation and development**

S. No.	Name of Ingredients	F1	F2	F3	F4	F5	F6
1	Hydroxychloroquine sulphate (mg)	100	100	100	100	100	100
2	Gelatin (mg)	200	250	300	350	400	450
3	PVP (mg)	200	300	350	250	450	400
4	Dibutyl Pthalate (%)	30	30	30	30	30	30



## 2.4 EVALUATION OF FORMULATION

### 2.4.1 Thickness of the patch:

The thickness of the drug loaded patch is measured in different points by using a digital micrometer and determines the average thickness and standard deviation for the same to ensure the thickness of the prepared patch<sup>5</sup>.

### 2.4.2 Uniformity of weight

Weights variation is studied by individually weighting randomly selected patches and calculating the average weight . the individual weight should not deviate significantly from the average weight<sup>5</sup>.

### 2.4.3 Folding endurance:

The patches were repeatedly folded at the same place till it broke. The number of times the patches could be folded at the same place without breaking gives the accurate value of folding endurance<sup>5</sup>.

### 2.4.4 Percentage Moisture content:

The prepared films are to be weighed individually and to be kept in a desiccator Containing fused calcium chloride at room temperature for 24 hrs. After 24 hrs the films are to be reweighed and determine the percentage moisture content from the below mentioned formula. Percentage moisture content =  $[\text{Initial weight} - \text{Final weight} / \text{Final weight}] \times 100^6$ .

### 2.4.5 Percentage Moisture uptake:

The weighed films are to be kept in a desiccator at room temperature for 24 hrs containing saturated solution of potassium chloride in order to maintain 84% RH. After 24 hrs the films are to be reweighed and determine the percentage moisture uptake from the below mentioned formula. Percentage moisture uptake =  $[\text{Final weight} - \text{Initial weight} / \text{initial weight}] \times 100^6$ .

### 2.4.6 Drug content :

A specified area of patch is to be dissolved in a suitable solvent in specific volume. Then the solution is to be filtered through a filter medium and analyze the drug contain with the suitable method (UV or HPLC technique). Each value represents average of three different samples<sup>6</sup>.



#### 2.4.7 *In vitro* skin permeation studies:

An *in vitro* permeation study can be using diffusion carried out by cell. Full thickness abdominal skin of male Wistar rats weighing of the diffusant. The temperature of the cell was maintained at  $32 \pm 0.5^{\circ}\text{C}$  using a thermostatically controlled heater. The isolated rat skin piece is to be mounted between the compartments of 200 to 250g. Hair from the abdominal region is to be removed carefully by using a electric clipper; the dermal side of the skin was thoroughly cleaned with distilled water to remove any adhering tissues or blood vessels, equilibrated for an hour in dissolution medium or phosphate buffer pH 7.4 before starting the experiment and was placed on a magnetic stirrer with a small magnetic needle for uniform distribution the diffusion cell, with the epidermis facing upward into the donor compartment. Sample volume of definite volume is to be removed from the receptor compartment at regular intervals, an equal volume of fresh medium is to be replaced. Samples are to be filtered through filtering medium and can be analyzed spectro photometrically<sup>7</sup>.

### 3. RESULTS AND DISCUSSION

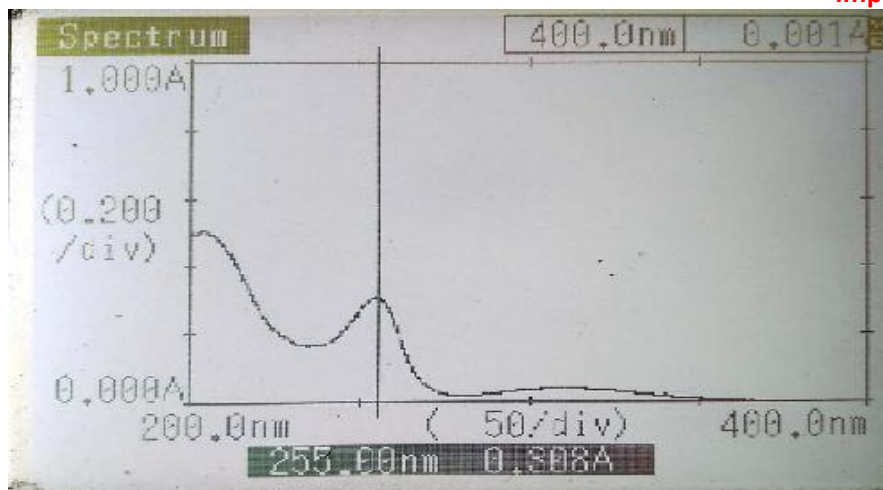
#### 3.1 Preformulation:

##### 3.1.1 Melting point determination:

The melting point of was found to be **Hydroxychloroquine sulphate**  $200^{\circ}\text{C}$  which is same as reported in literature.

##### 3.1.2 Determination of wavelength using UV spectrophotometric analysis:

The maximum wavelength of was found **Hydroxychloroquine sulphate** to be 255 nm which matches the reported wavelength.



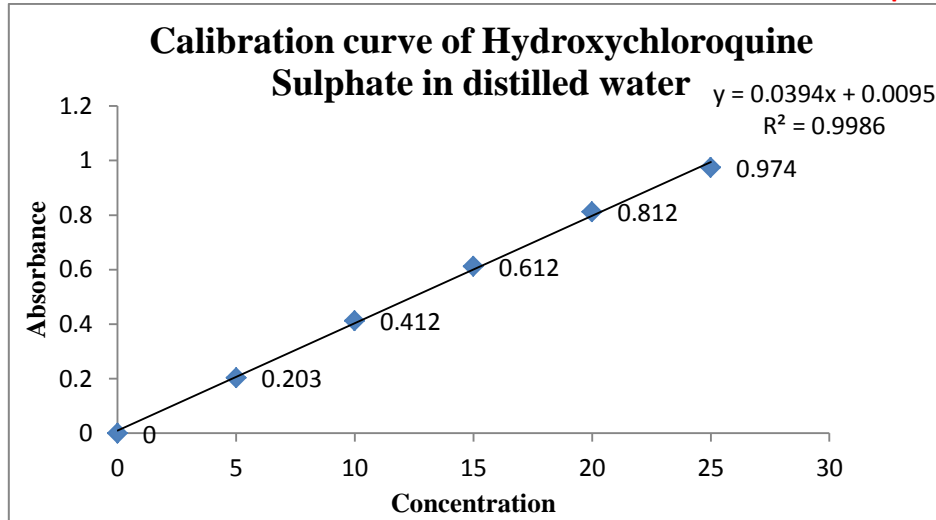
**Figure-2 : UV spectrum of Hydroxychloroquine sulphate**

### 3.1.3 Preparation of calibration curves:

The calibration curves of Hydroxychloroquine sulphate in various solvents e.g distilled water, 6.8 pH phosphate buffer, 7.4 pH phosphate buffers were prepared and shown below:

**Table No. 3: Absorbance data of Hydroxychloroquine sulphate in distilled water for preparation of calibration curve, at 255 nm**

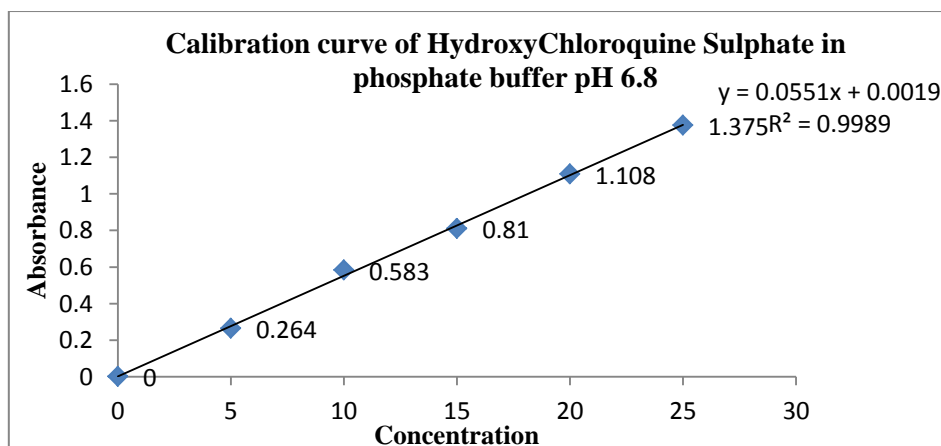
S.No	Concentration $\mu\text{g/ml}$	Absorbance Mean $\pm$ Standard Deviation
01	05	0.203
02	10	0.412
03	15	0.612
04	20	0.812
05	25	0.974



**Figure-4** Calibration curve of Hydroxychloroquine sulphate in distilled water

**Table No. 5:** Absorbance data of Hydroxychloroquine sulphate in phosphate buffer pH 6.8 for preparation of calibration curve, at 255 nm

S.No	Concentration µg/ml	Absorbance Mean Standard Deviation
01	05	0.264
02	10	0.583
03	15	0.810
04	20	1.108
05	25	1.375

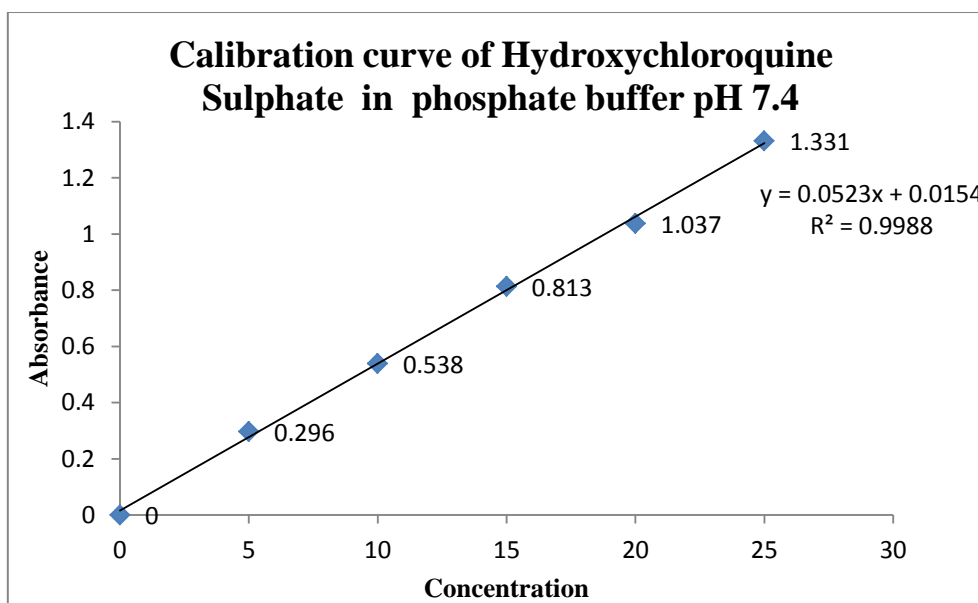


**Figure- 6** Calibration curve of Hydroxychloroquine sulphate in phosphate buffer pH 6.8



**Table No. 7: Absorbance data of Hydroxychloroquine sulphate in phosphate buffer 7.4 for preparation of calibration curve**

S.No	Concentration $\mu\text{g/ml}$	Absorbance Mean $\pm$ Standard Deviation
01	05	0.296
02	10	0.538
03	15	0.813
04	20	1.037
05	25	1.331



**Figure- 8 Calibration curve of Hydroxychloroquine sulphate in phosphate buffer pH 7.4**

### 3.1.4 Determination of Partition coefficient:

The partition coefficient was found to be  $\log P = 4.63$ . The partition coefficient found cleared that the drug is highly lipophilic.

### 3.1.5 Determination of solubility of Hydroxychloroquine sulphate in various medium:

The solubility of hydroxychloroquine sulphate in various mediums was studied and the results of study were shown in below table:

**Table-9: Solubility data of Hydroxychloroquine sulphate in different mediums**

S.NO	Solvent	Solubility(mg/ml) Mean $\pm$ SD	Inference
01	Distilled water	300 $\pm$ 1 ml	Practically soluble
02	Phosphate buffer pH 6.8	280 $\pm$ 1 ml	Practically soluble
03	Phosphate buffer pH 7.4	290 $\pm$ 1 ml	Practically soluble

### 3.1.6 Drug-excipient interaction study

The drug Hydroxychloroquine sulphate was found to be compatible with various excipients which were selected for formulation of TDDS. The compatibility was assessed by TLC and the retention factors of all ratios found similar.

**Table-10: Data of drug-excipient interaction study**

S.No	Drug/Drug + Excipient Ratio (1:1)	Initial appearance	Final appearance	Retention factor
01	Drug (Hydroxychloroquine phosphate)	White powder	White powder	0.55
02	Pure Drug + Gelatin	Light yellow	Light yellow	0.56
03	Pure Drug + Dibutyl phthalae	White transparent	White transparent	0.53
04	Pure Drug + Polyvinyl pyrrolidone	White	White	0.58



## **3.2 Evaluation of Transdermal Patch:**

### **3.2.1 Folding Endurance**

Folding endurance is determined to identify flexibility of the films. Folding endurance was determined by folding the films repeatedly in the same part of the film until it broke.

### **3.2.2 Thickness**

Five films from each formulation were selected randomly to study the thickness using thickness gauge and average was determined.

### **3.2.3 Weight Variation Test**

The study was carried out by determining weight of randomly selected five films from each batch with the help of high accuracy electronic balance. The average weight of a film and its standard deviation was calculated.

### **3.2.4 Percentage of Moisture Content**

Randomly selected films were weighed individually and kept in the platform of the dessicator containing anhydrous calcium chloride at room temperature for 24 hours. Films were weighed separately and repeatedly until a constant weight obtained. The percentage of moisture content was calculated by the difference between initial and final weight with respect to the final weight.

### **3.2.5 Drug Content Determination:**

Selected film from each batch was put into a 100 ml standard flask containing the buffer solution (pH - 7.4) and shaken continuously for 24 hours. Then the solution was filtered and drug content was determined with the help of UV spectrophotometer at wave m. length 255 nm.

**Table-11: Data of Folding Endurance Thickness, Weight Variation, % Moisture content, % Moisture Uptake, Drug content.**

Formulation Code	Folding Endurance	Thickness ( $\mu\text{m}$ )	Weight Variation (mg)	% Moisture Content	% Moisture Uptake	Drug Content
F1	98 $\pm$ 4.28	160 $\pm$ 3.03	10.6 $\pm$ 0.37	2.32 $\pm$ 0.56	1.83 $\pm$ 0.42	97.9 $\pm$ 2.42
F2	102 $\pm$ 4.02	168 $\pm$ 5.88	11.12 $\pm$ 0.26	2.92 $\pm$ 0.68	2.63 $\pm$ 0.56	98.6 $\pm$ 2.45
F3	96 $\pm$ 6.03	170 $\pm$ 5.95	10.23 $\pm$ 0.36	4.02 $\pm$ 0.89	3.63 $\pm$ 0.35	97.5 $\pm$ 2.41
F4	110 $\pm$ 5.22	158 $\pm$ 5.53	11.20 $\pm$ 0.29	1.96 $\pm$ 0.39	1.32 $\pm$ 0.31	96.9 $\pm$ 2.39
F5	103 $\pm$ 5.00	166 $\pm$ 5.81	10.73 $\pm$ 0.35	1.78 $\pm$ 0.33	1.25 $\pm$ 0.26	98.8 $\pm$ 2.45
F6	108 $\pm$ 5.32	154 $\pm$ 5.39	10.97 $\pm$ 0.40	1.64 $\pm$ 0.31	1.12 $\pm$ 0.34	95.8 $\pm$ 2.35

### 3.2.7 IN VITRO DRUG PERMEATION STUDIES

*In vitro* permeation studies were performed using Franz diffusion cell. The dialysis membrane (2x2cm) was mounted between the donor and receptor compartment of the diffusion cell. The dialysis sac was previously soaked for 2 h in PBS. The formulated films were placed over the membrane. The receptor compartment of the diffusion cell was filled with phosphate buffer pH 7.4. The solution in the receptor compartment was constantly and continuously stirred using magnetic bead at 50 rpm; the temperature was maintained at 37  $\pm$  20C. Samples were withdrawn (2 ml) at predetermined time intervals and replaced with an equal volume of phosphate buffer. The samples were suitably diluted and analyzed to determine drug content using UV spectrophotometer at a wave length of 255nm.

## 4. SUMMARY AND CONCLUSION

Transdermal therapeutic systems are high- tech patches that make treatment much more convenient and pleasant for patient .Instead of having to take lots of tablets, patients often only have to apply a new patch once a week In addition, patches from LTS also ensure a constant drug level for the entire duration of wear Symptoms upon getting up in the morning as a result of low drug levels during the night are avoided. There are five major advantages of transdermal patches above other systems like Improved well –being for patient, constant



drug levels, fewer side effects, Drugs that cannot be absorbed via the gastrointestinal tract can now be used and because the substance is released evenly and directly into the blood stream, less of the substance may be required.

In the application of delivery systems for the delivery of Hydroxychloroquine Sulphate is the design of the systems affects their therapeutic efficacy. In the design of polymer – drug conjugates for the delivery of Hydroxychloroquine Sulphate is the nature of targeting moieties affect the specificity of the conjugates , materials used influence their degree of toxicity, the type of drug linkers determine the drug release mechanism from the conjugates and the position of functionalities on the conjugates affect their rate of degradation. The design of micells influence their degree of toxicity, rate of biodegradation and the rate of diffusion of the incorporated drugs. However, some of these delivery systems are limited by aggregation upon storage, burst drug release effects and susceptibility of selected drug linkers to some enzymes resulting in rapid drug release . There are few reports on the delivery of Hydroxychloroquine Sulphate is to form selected systems such as ethosomes, nanocapsules niosomes and CNTs indicating that there is a need for further investigations on these systems. Most of the systems were evaluated in vitro and in vivo and the results obtained were promising suggesting that there is a pressing need for further studies to be performed on these systems to reach clinical trials because emergence of drug resistance remains a global problem.

## 5. ACKNOWLEDGEMENT

I am thankful to the management of School of Pharmacy, Dr. APJ Abdul Kalam University Indore. For providing necessary facilities to carry out the research work and heartily thankful to my guide Dr. Rakesh Patel for providing all the support and encouragement to carry out this studies.



Swati Yadav *et al*, Int. Journal of Pharmaceutical Sciences and Medicine (IJPSM),  
Vol.8 Issue. 7, July- 2023, pg. 67-80

ISSN: 2519-9889

Impact Factor: 5.9

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